

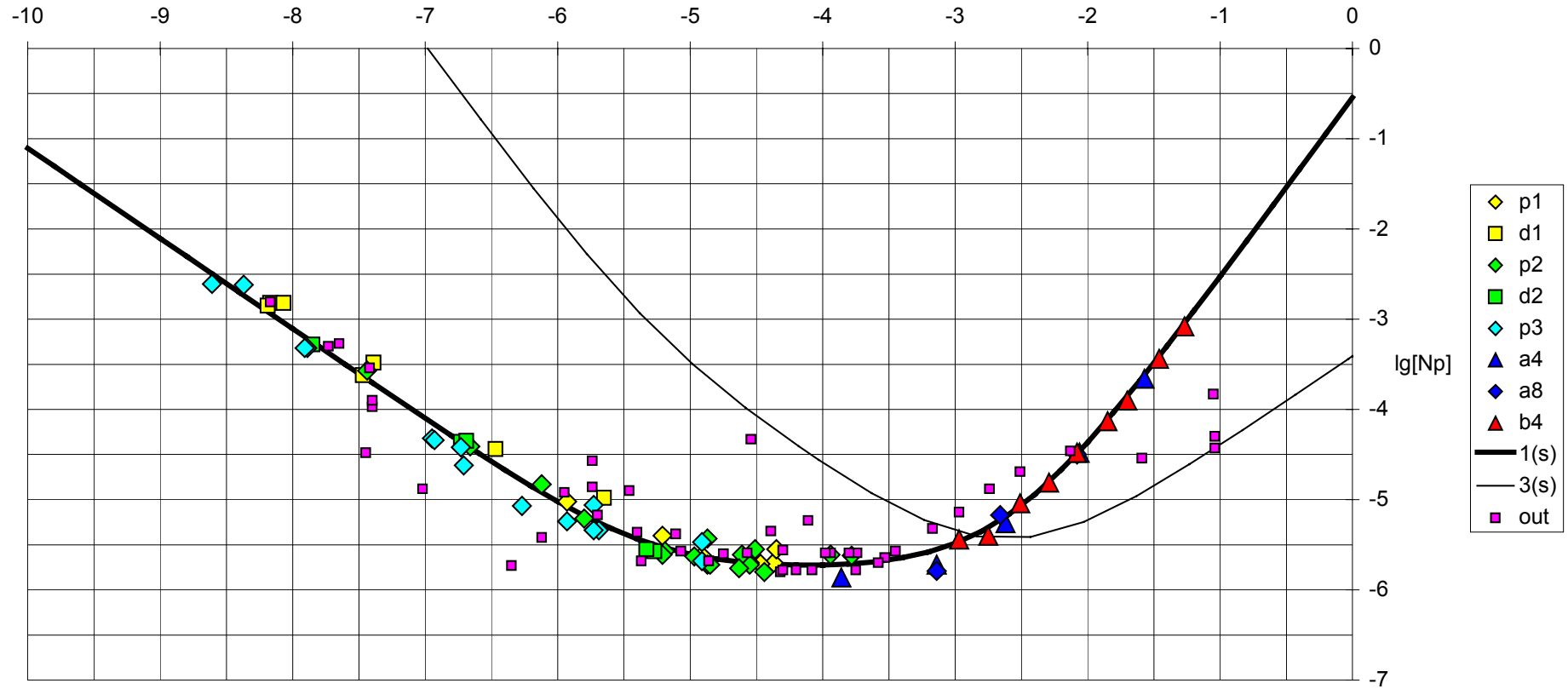
Figures: pierre.vitorge(at)cea.fr

Treatment of Np(V) solubility measurements  
in carbonate / bicarbonate media.

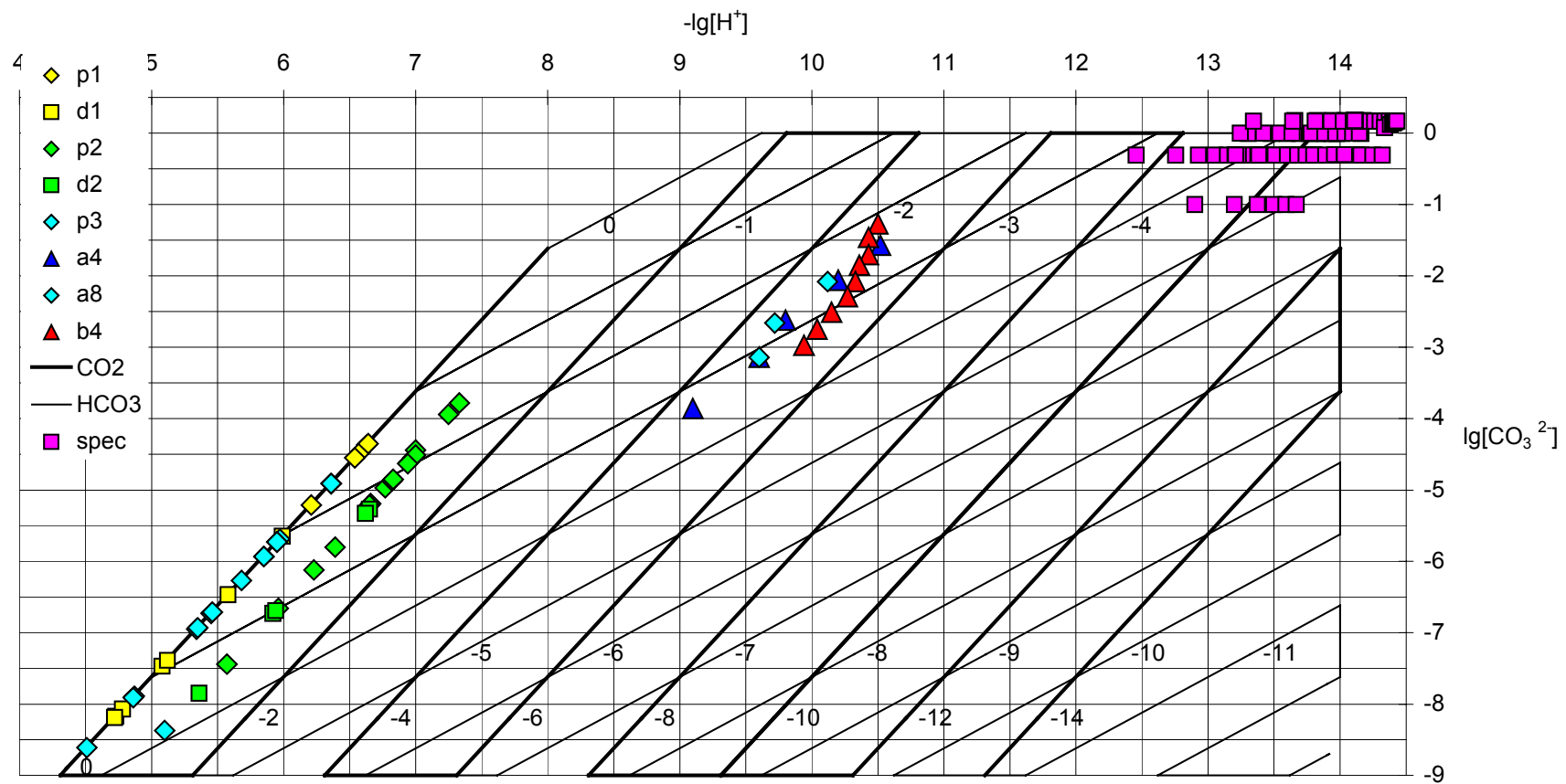
Room temperature (21°C), 3 M NaClO<sub>4</sub> aqueous solutions.

CO<sub>2</sub>(g) bubbling in HCO<sub>3</sub><sup>-</sup> aqueous solution,  
or bicarbonate or carbonate batches.

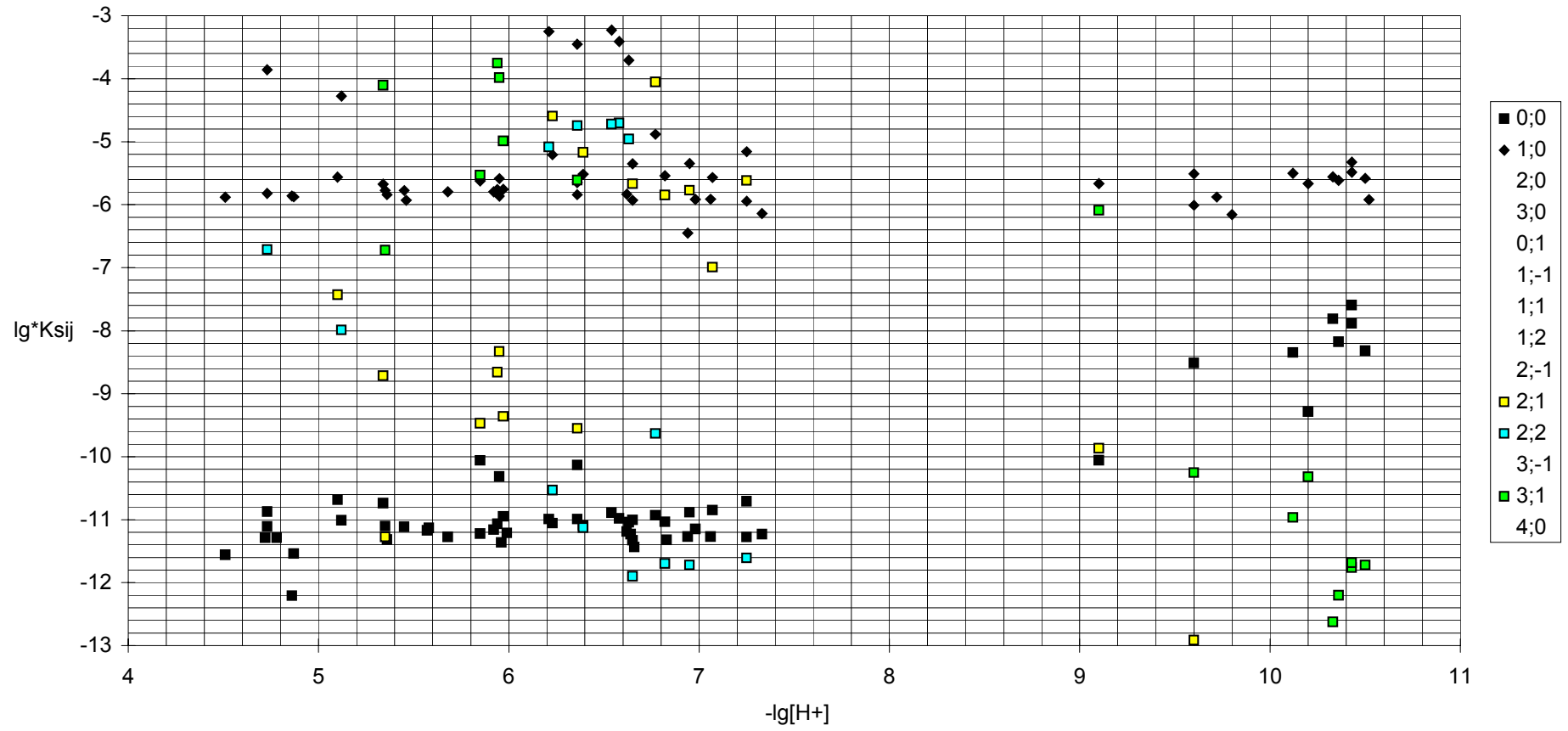
### NaNpO<sub>2</sub>CO<sub>3</sub> solubility



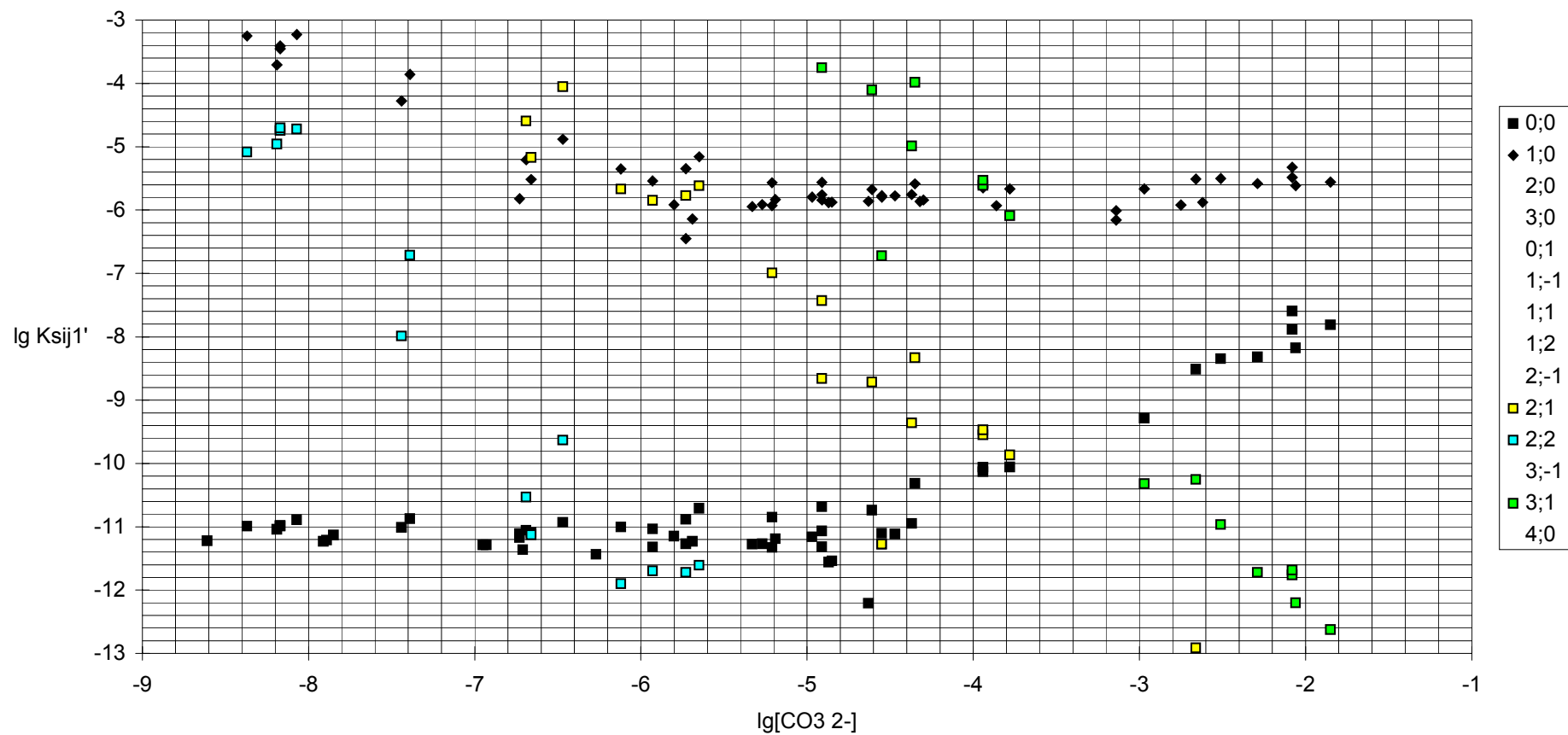
### Experimental domain for Np(V) solubility measurements



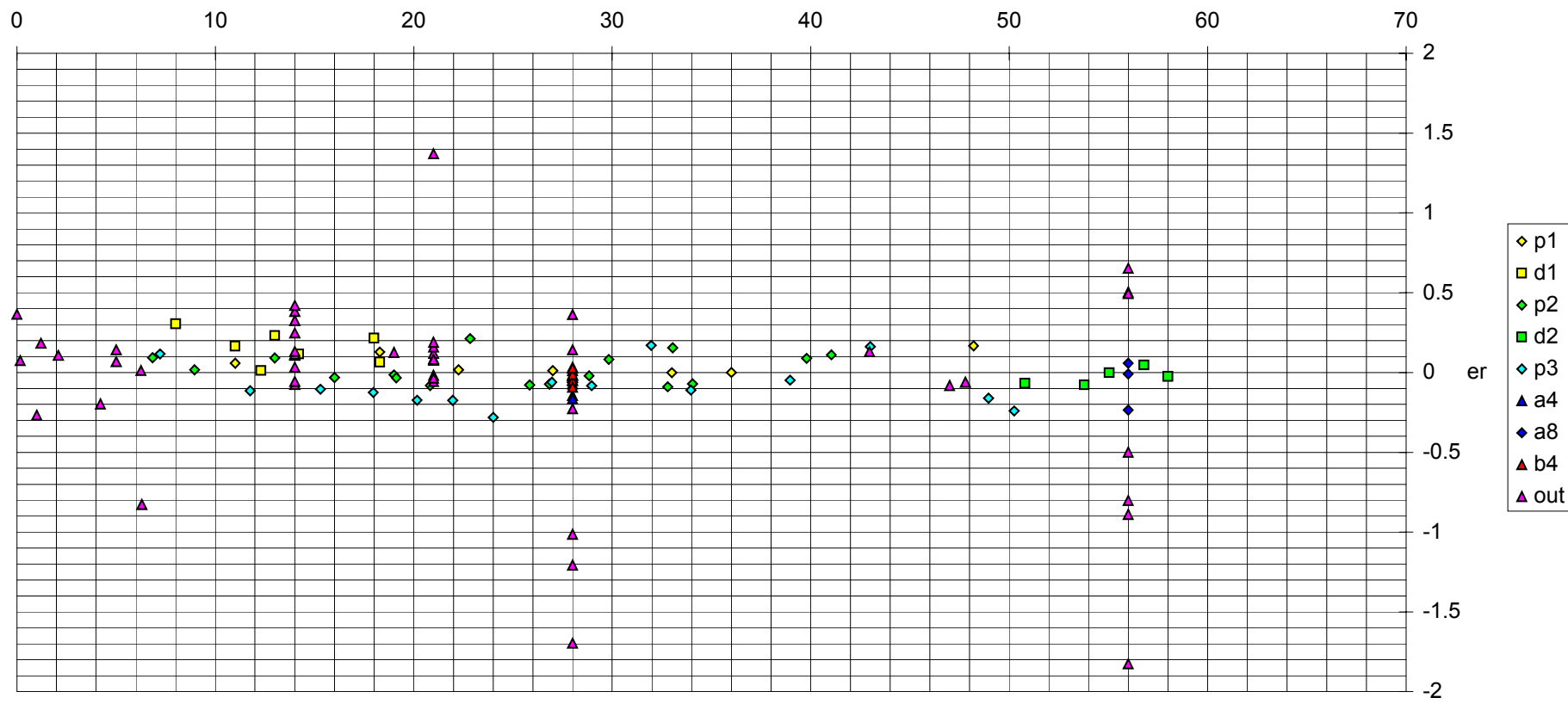
Ig\*Ksij results versus Ig[H+]



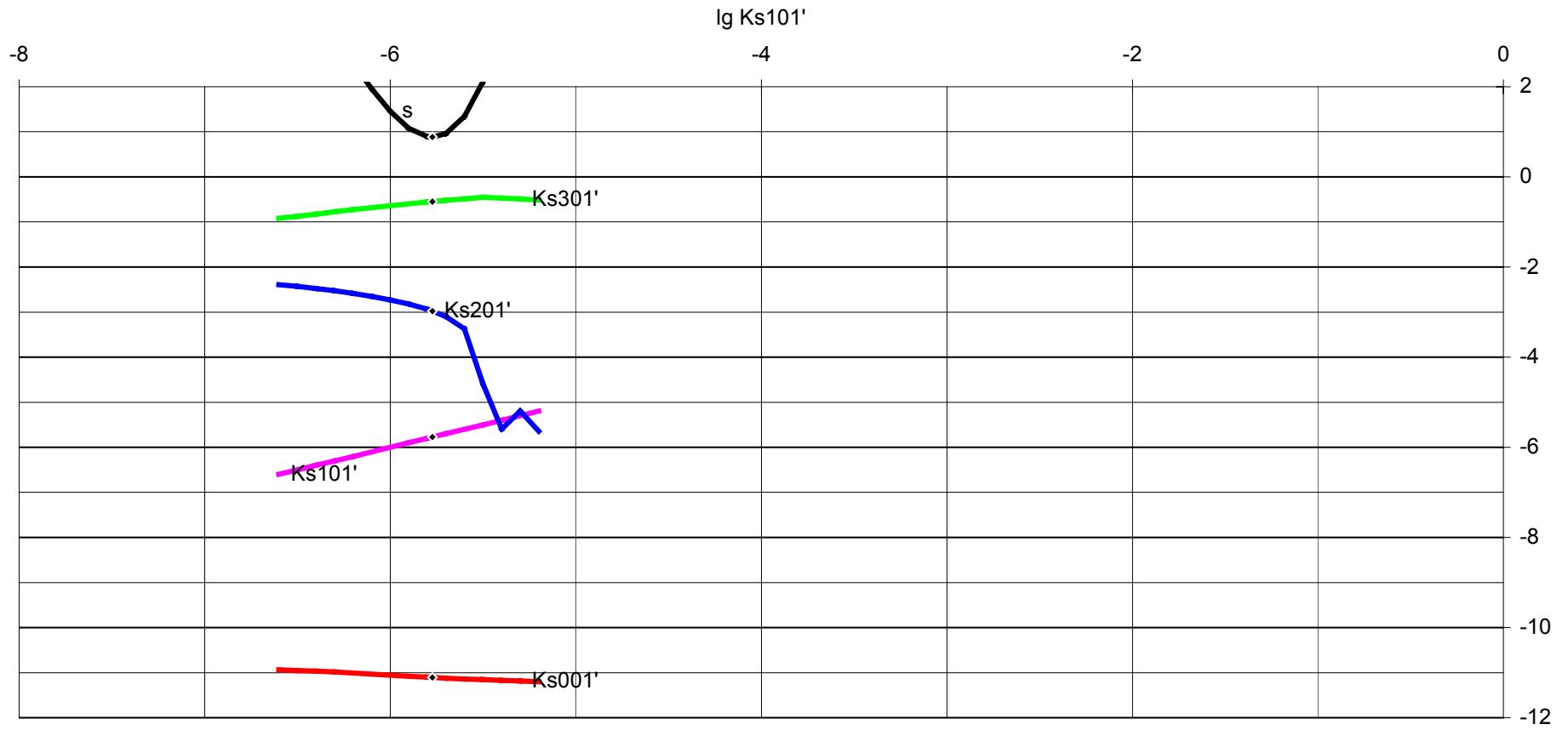
Ksij1' [CO3] dependency



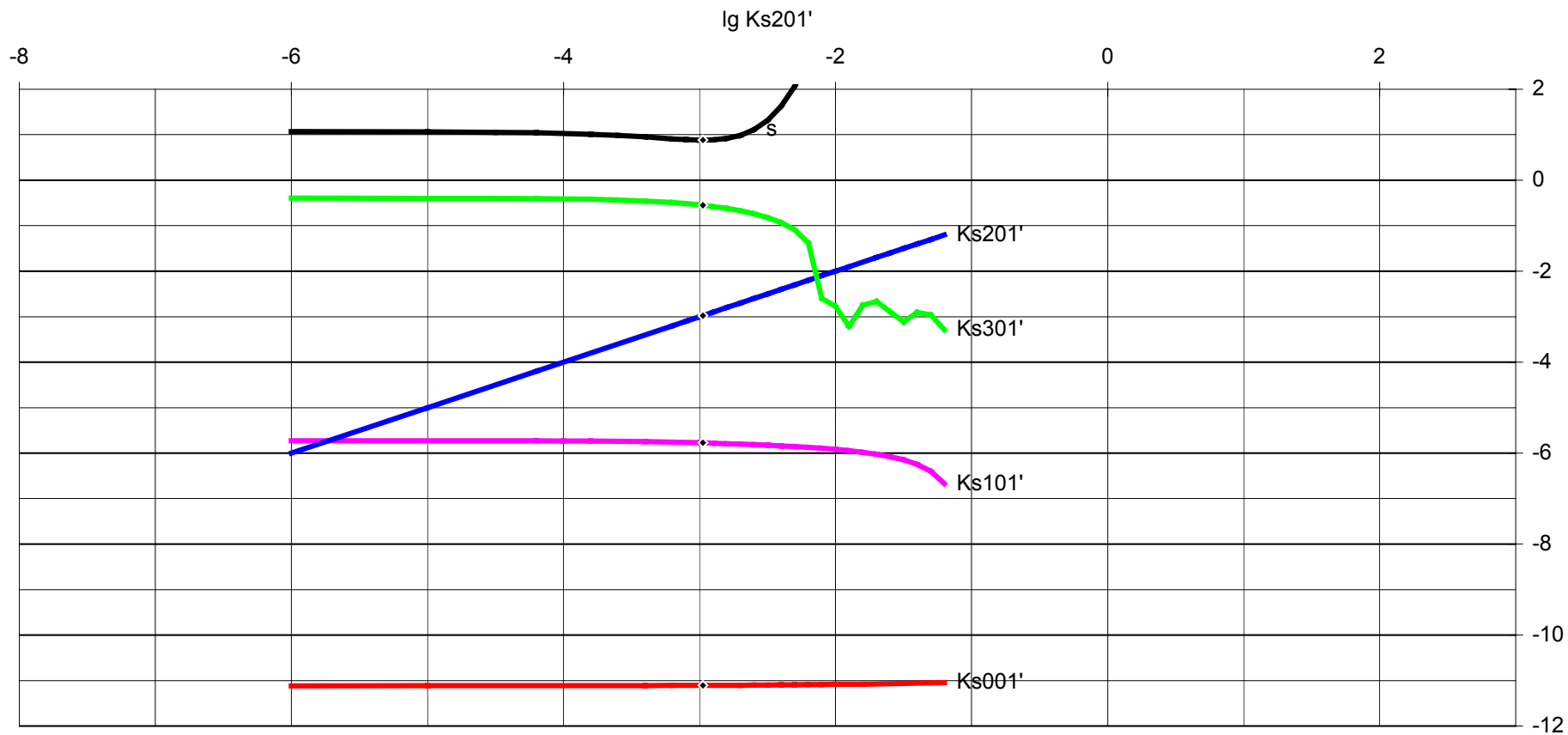
# NaNpO2CO3 solubility



### NpO<sub>2</sub>CO<sub>3</sub>- formation constant dependency of curve fitting results

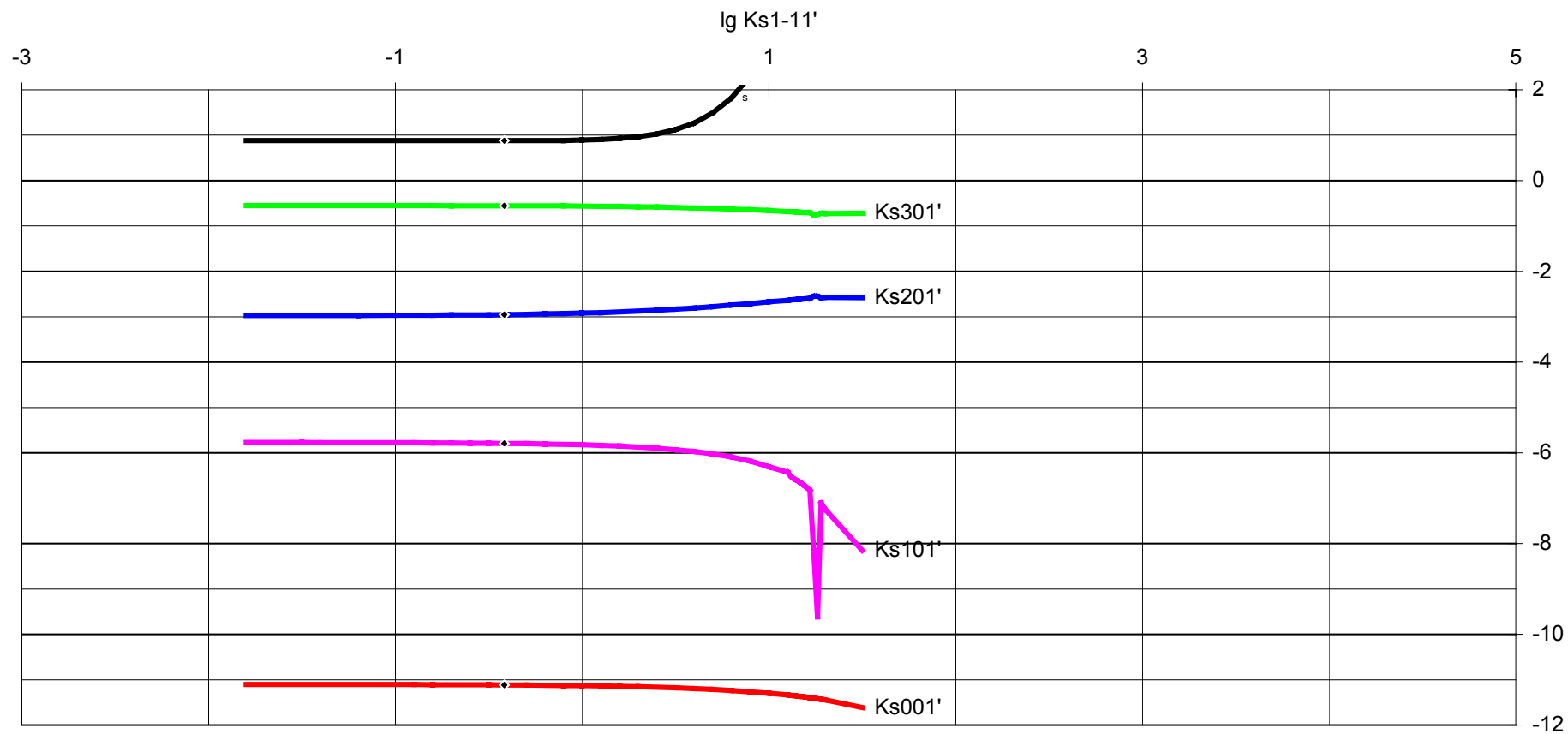


### NpO<sub>2</sub>(CO<sub>3</sub>)<sub>2</sub> 3- formation constant dependency of curve fitting results

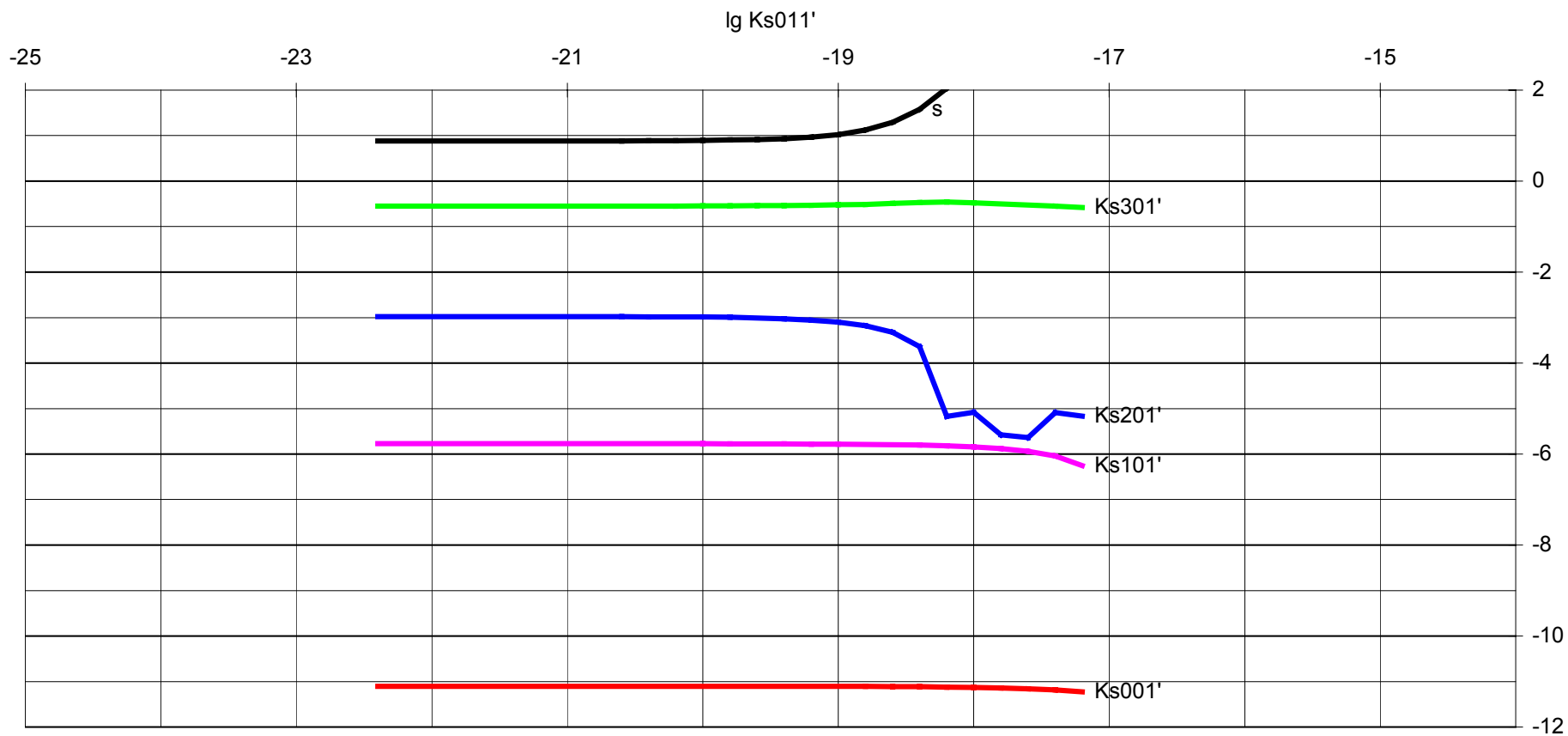




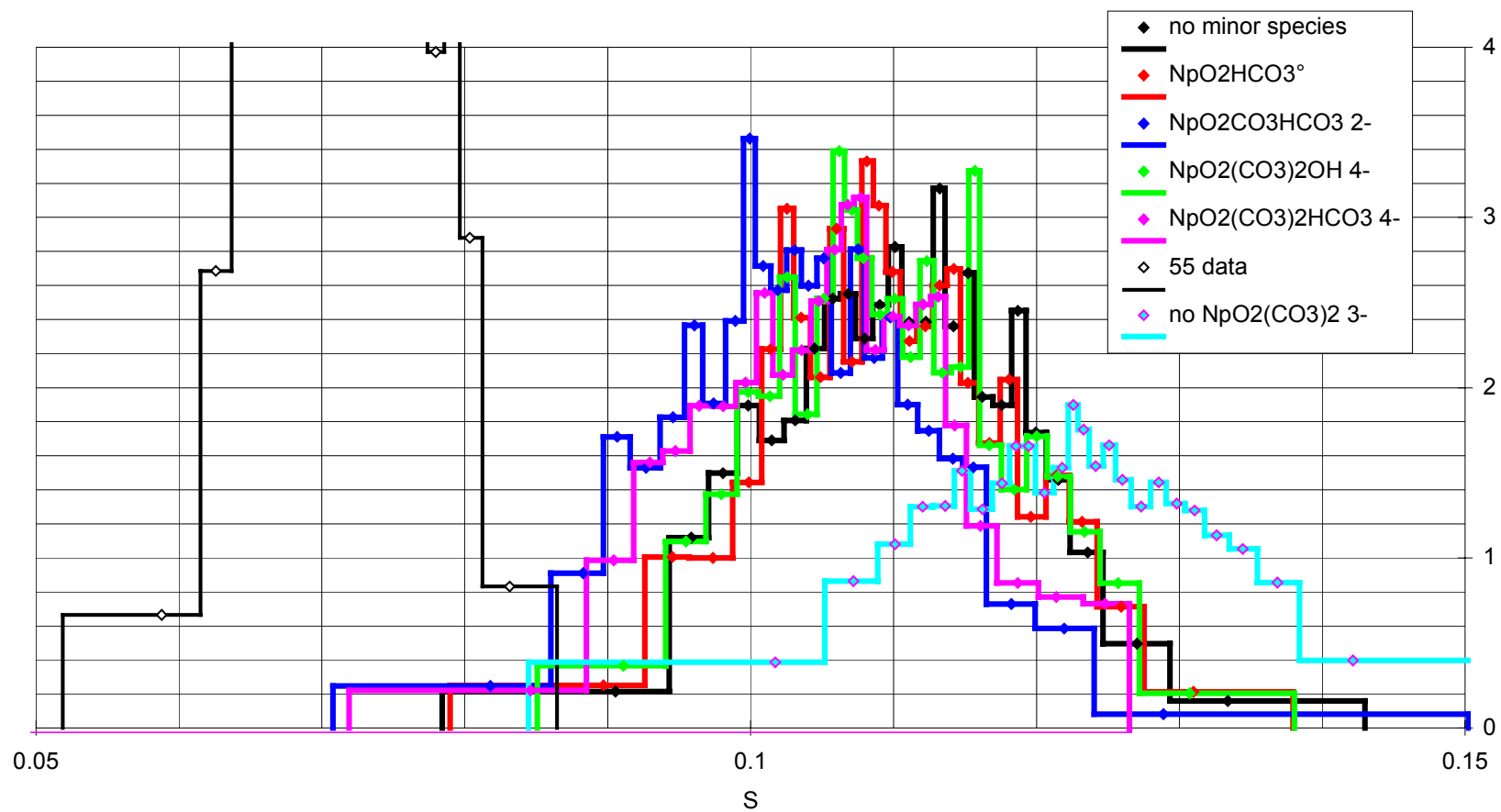
### NpO<sub>2</sub>HCO<sub>3</sub><sup>o</sup> formation constant dependency of curve fitting results



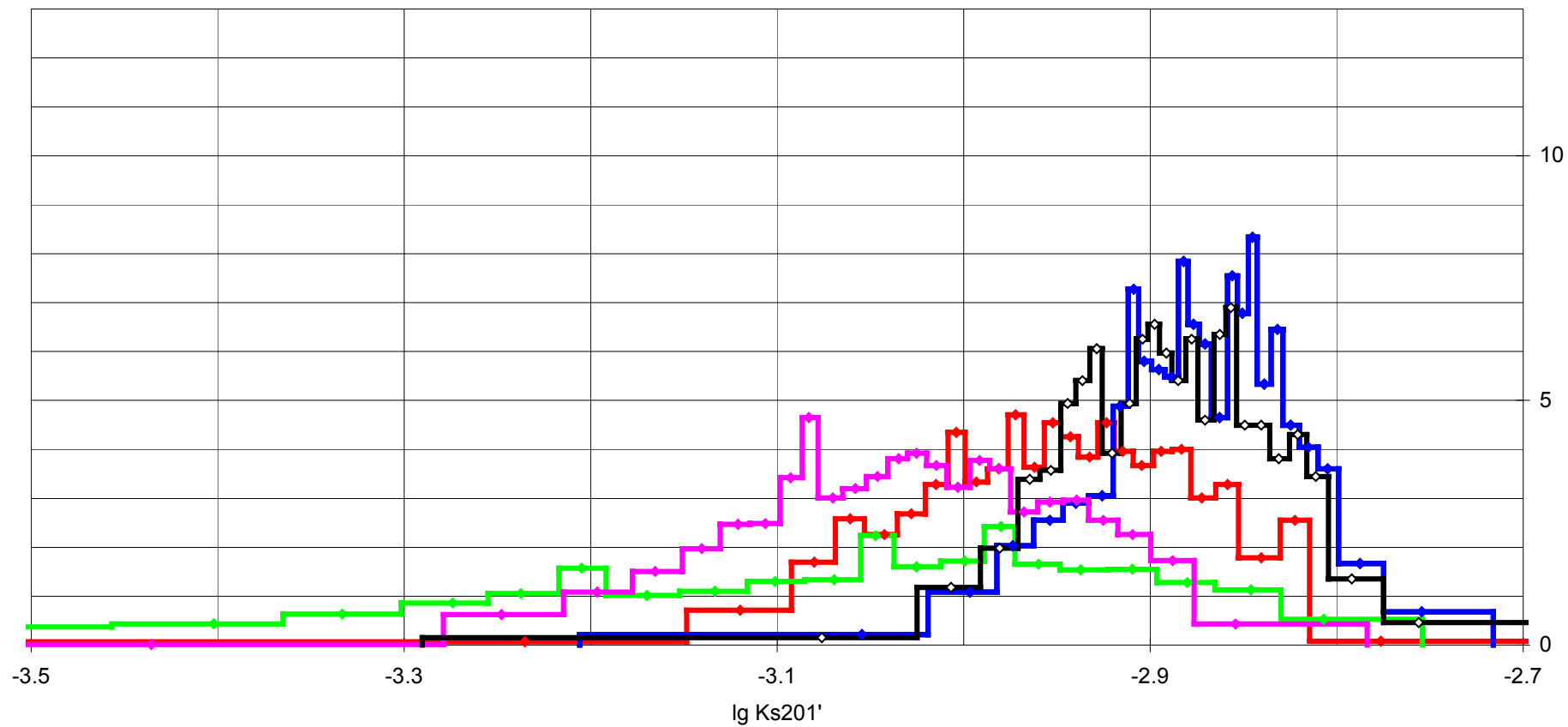
NpO<sub>2</sub>OH° formation constant dependency of curve fitting results (On Np(V) solubility data in HCO<sub>3</sub><sup>-</sup>/CO<sub>3</sub><sup>2-</sup> media)



Minor species influence on the deviation error distribution.



Minor species influence on the lg Ks201' error distribution (Np(V) solubility data).

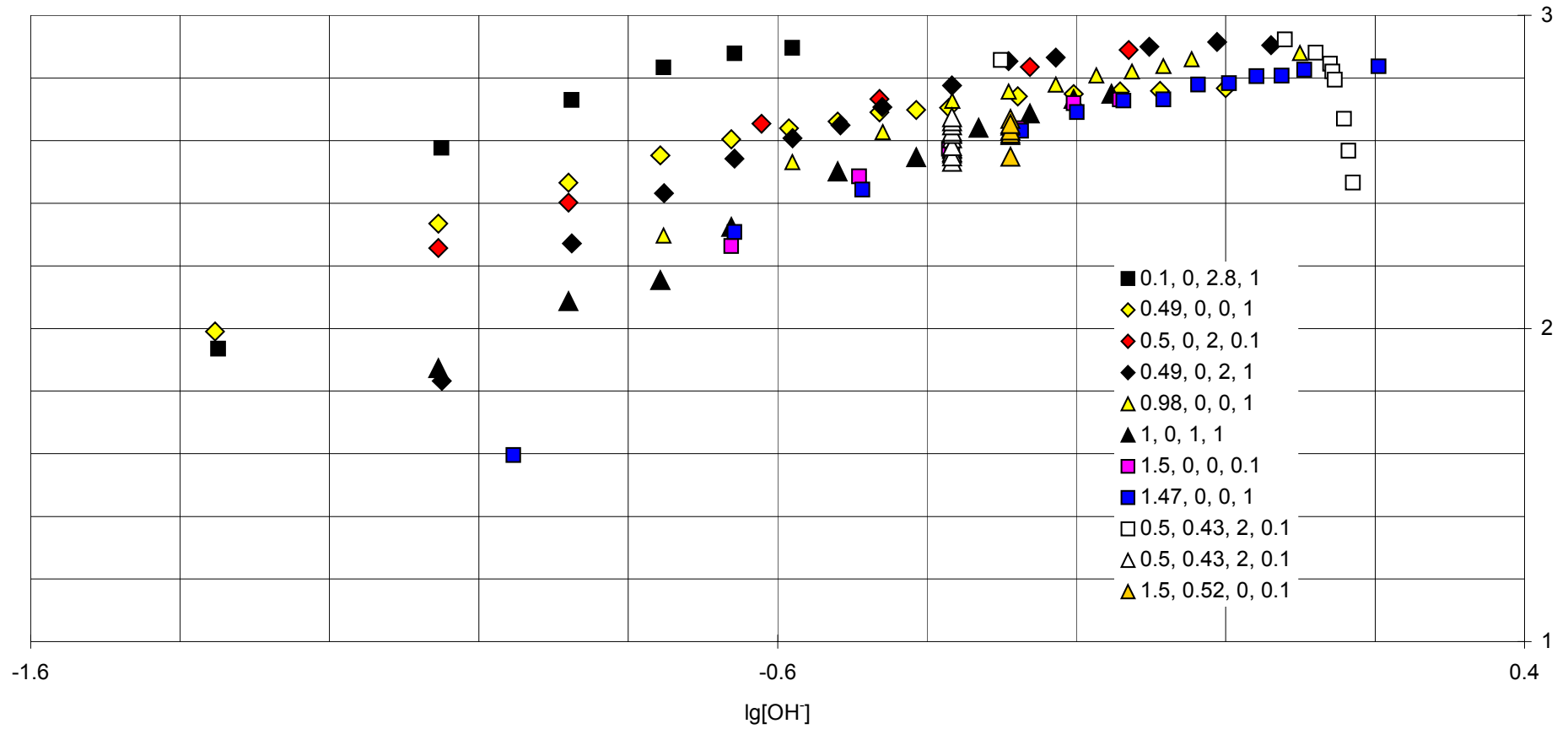


Figures:

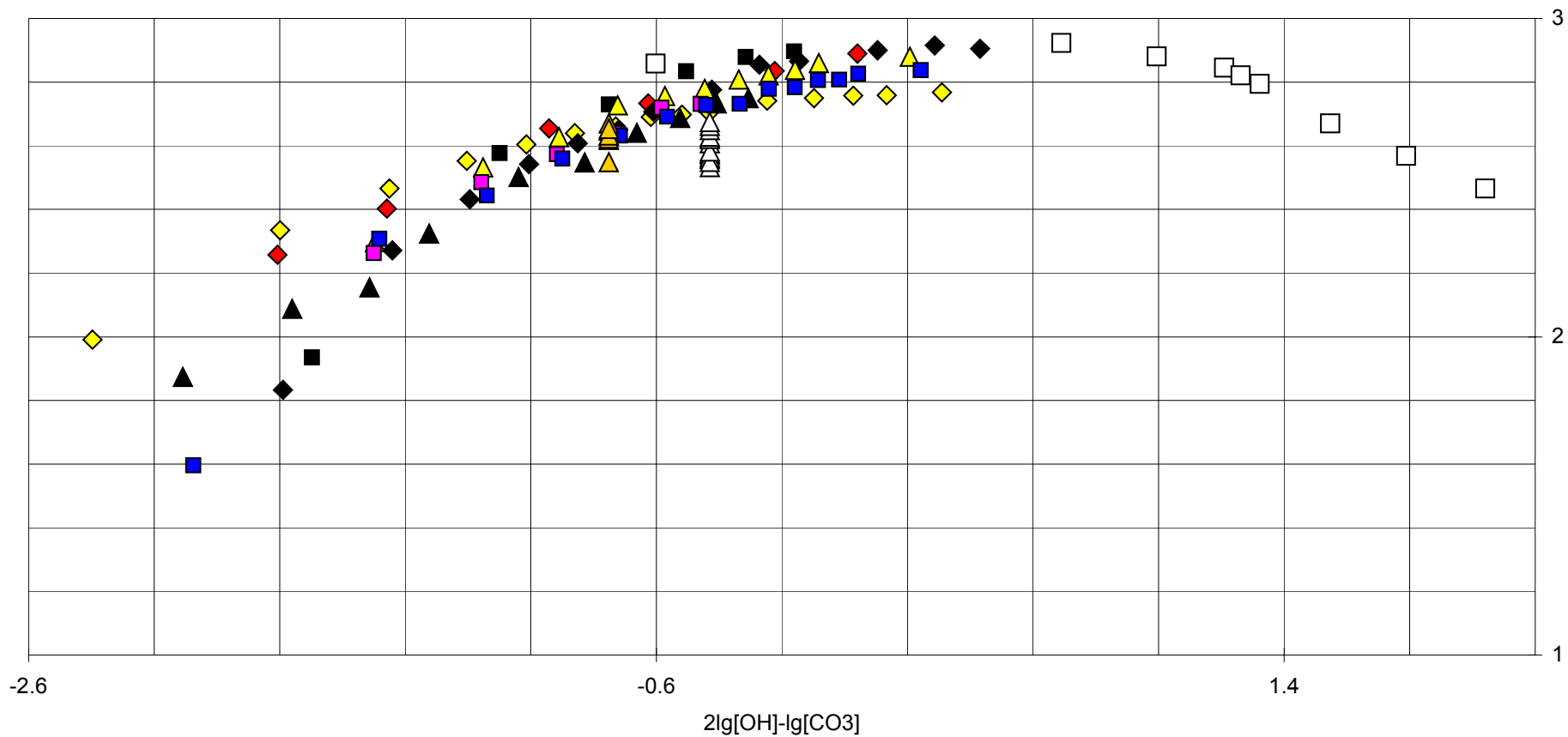
Treatment of Np(V) spectrophotometric measurements  
in carbonate / hydroxide media.

Room temperature (21°C), 3 M NaClO<sub>4</sub> aqueous solutions.  
Concentrated NaOH aqueous solutions were directly added in the  
spectrophotometric (10 cm path length) cuvette previously filled  
with a Na<sub>2</sub>CO<sub>3</sub> aqueous solution.

Np(V) absorbance in carbonate hydroxide media



Np(V) aborbance in carbonate hydroxide media



Np(V) in carbonate hydroxide media (experimental domain)

